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Impulse J produced from time t_1 to t_2 is defined to be $= \int_{t_1}^{t_2} F dt$ where F is the resultant force applied from t_1 to t_2 . From Newton's second law, force is related to momentum p by $F = dp/dt$. Therefore, $J = \int_{t_1}^{t_2} dp = \Delta p$ where Δp is the change in linear momentum from time t_1 to t_2 . This is often called the impulse-momentum theorem. As a result, an impulse may also be regarded as the change in ...

Impulse (physics) - Wikipedia

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The popular theory of atomic structure at the time of Rutherford's experiment was the "plum pudding model". This model was devised by Lord Kelvin and further developed by J. J. Thomson. Thomson was the scientist who discovered the electron, and that it was a component of every atom. Thomson believed the atom was a sphere of positive charge throughout which the electrons were distributed, a bit ...

Geiger–Marsden experiment - Wikipedia

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Solucionarios en Ingles

Photoelectric Effect
1887
1905

Photoelectric Effect - Wikipedia

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Photoelectric Effect - Wikipedia

En un artículo de 1909, «En una Reflexión Difusa de las Partículas Alfa», [6] Geiger y Marsden describieron el experimento mediante el cual demostraron que las partículas alfa pueden ser dispersadas por más de 90°. En su experimento prepararon un

